

## Label-free Rapid Semiquantitative Detection of Proteins Down to Sub-monolayer Coverage by Using Surface-enhanced Raman Scattering of Nitrate Ion

Zhen Zhou,<sup>1</sup> Genin Gary Huang,<sup>1,2</sup> and Yukihiro Ozaki\*<sup>1</sup>

<sup>1</sup>Department of Chemistry, Kwansai Gakuin University, Sanda, Hyogo 669-1337

<sup>2</sup>Department of Medicinal and Applied Chemistry, Kaohsiung Medical University, Kaohsiung 807, Taiwan

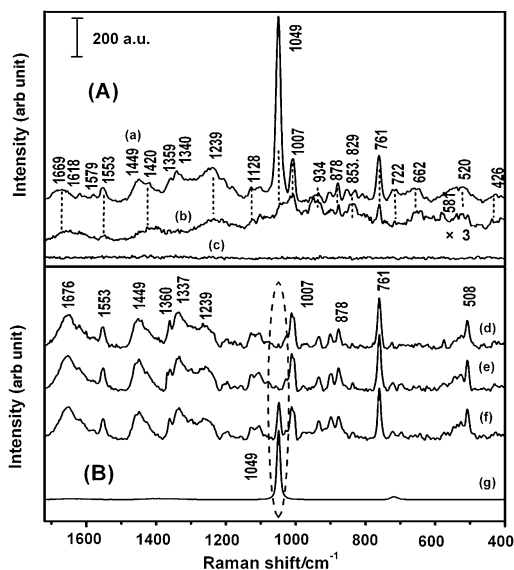
(Received June 28, 2010; CL-100588; E-mail: ozaki@kwansai.ac.jp)

A novel heat-induced SERS method was used to selectively enhance a band of  $\text{NO}_3^-$  at  $1049\text{ cm}^{-1}$  for semi-quantitative detection of lysozyme and insulin down to  $10^{-9}$  and  $10^{-8}\text{ M}$ , respectively, and bell shape variations of SERS intensities were observed for the concentration dependency of the proteins.

Surface-enhanced Raman scattering (SERS) has recently proved its potential in ultra-sensitive detections of biomolecules.<sup>1–6</sup> However, despite a great number of studies done in this field, label-free detections of proteins which do not contain resonant chromophores, still remain a challenge. Although label-free non-resonant SERS spectra have been obtained for several kinds of proteins,<sup>7–12</sup> no accurate quantitative study of proteins by nonresonant SERS has been reported. To our best knowledge, the only one relative quantitative study was realized by our group.<sup>6</sup> Because proteins without resonant chromophores have, in general, low Raman cross sections, the progress in label-free detections of this class of proteins is rather slow. The major advantages of the method used in the present study exist in its high sensitivity and routine analyses for semi-quantitative study.

We have recently proposed a heat-induced SERS-sensing method for rapid detection of a peptide, glutathione.<sup>13</sup> In this method,  $30\ \mu\text{L}$  of glutathione solution was heated on an aluminum pan plate at  $100\text{ }^\circ\text{C}$  for 3 min to obtain a dry film of glutathione, and then its SERS signal was acquired. Comparing to conventional SERS methods, this protocol enables one to obtain much larger SERS enhancement without loss of vibrational information about an analyte. In the present study, the application of this heat-induced SERS method is extended to protein detection. In the present study,  $\text{NaNO}_3$  is used as an electrolyte instead of the most commonly used hydrochloride salt.  $\text{Cl}^-$  binds tightly to silver colloid<sup>4</sup> and thus it is difficult for an analyte to competitively adsorb on the silver surface to yield strong SERS signal.<sup>4,6</sup> First, a citrate buffer containing 6 mM  $\text{NaNO}_3$  was mixed with a Ag colloid solution. Then, a protein was added to the mixture followed by drying the mixture at  $100\text{ }^\circ\text{C}$  for SERS measurement. Thanks to this highly sensitive method with combination of large Raman cross section of  $\text{NO}_3^-$ , semi-quantitative detection of proteins can be achieved down to  $10^{-9}\text{ M}$  for lysozyme and  $10^{-8}\text{ M}$  for insulin, respectively.

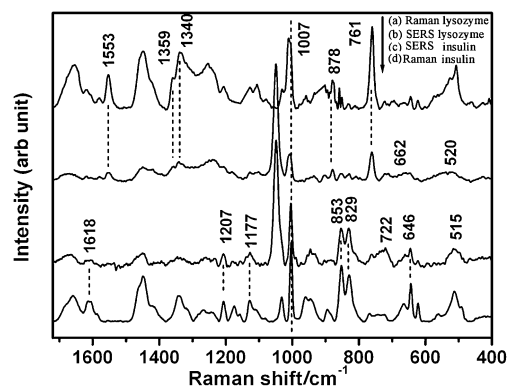
Figures 1a and 1b compare SERS spectra of lysozyme ( $10^{-5}\text{ M}$  in citrate-reduced silver colloid) measured by use of the heat-induced method and without heat treatment (solution sample). As control in Figure 1c, a normal Raman spectrum of a dry film of  $10^{-5}\text{ M}$  lysozyme solution is shown. In the SERS spectrum of lysozyme without the heat treatment (its intensity has been magnified by a factor of 3, Figure 1b), there appears several bands due to Trp, Tyr, Phe, disulfide groups ( $\nu(\text{C}-\text{S})$  and  $\nu(\text{S}-\text{S})$ ), amide I and amide III modes (see Table 1 in Supporting



**Figure 1.** (A)(a) A SERS spectrum of a dry film of the solution containing  $10^{-5}\text{ M}$  lysozyme and 6 mM  $\text{NaNO}_3$ , (b) a SERS spectrum of a solution containing  $10^{-5}\text{ M}$  lysozyme and 6 mM  $\text{NaNO}_3$ , and (c) a normal Raman spectrum of a dry film of  $10^{-5}\text{ M}$  lysozyme solution. (B)(d) A normal Raman spectrum of  $7 \times 10^{-3}\text{ M}$  lysozyme in pure water; (e), (f) normal Raman spectra of  $7 \times 10^{-3}\text{ M}$  lysozyme in a 5 mM citrate buffer (pH 4.0) containing 0 mM  $\text{NaNO}_3$  and 40 mM  $\text{NaNO}_3$ ; (g) a normal Raman spectrum of 1 M  $\text{NaNO}_3$ . The intensity bar is applied only for (A) while band intensities in (B) have been adjusted artificially for display.

Information (SI)<sup>22</sup> for detailed band assignments). In the heat-induced SERS spectrum, an intense peak at  $1049\text{ cm}^{-1}$  and a series of weaker bands are observed. In normal Raman spectra of solution lysozyme shown in Figures 1d and 1e, there is no peak at  $1049\text{ cm}^{-1}$  and this peak also cannot match any bands of citrate anion.<sup>14</sup> The only remaining possibility for the origin of the  $1049\text{ cm}^{-1}$  band is  $\text{NO}_3^-$ . To confirm this speculation, normal Raman spectra of lysozyme solutions containing 0 and 40 mM  $\text{NaNO}_3$  are compared (Figures 1e and 1f), and the band is found at  $1049\text{ cm}^{-1}$ , revealing that it arises from the  $\text{NO}_3^-$  vibration. This peak has seldom been reported in SERS references; however, Mrozek et al. unambiguously assigned it to the  $\text{NO}_3^-$  mode in their study of saccharide using electrochemically roughed silver surface.<sup>15</sup> Both their and our work suggest that the dry film methods selectively enhance peak of  $\text{NO}_3^-$  at  $1049\text{ cm}^{-1}$ .

SERS spectra of lysozyme and insulin obtained by the heat-induced method as well as their corresponding normal Raman

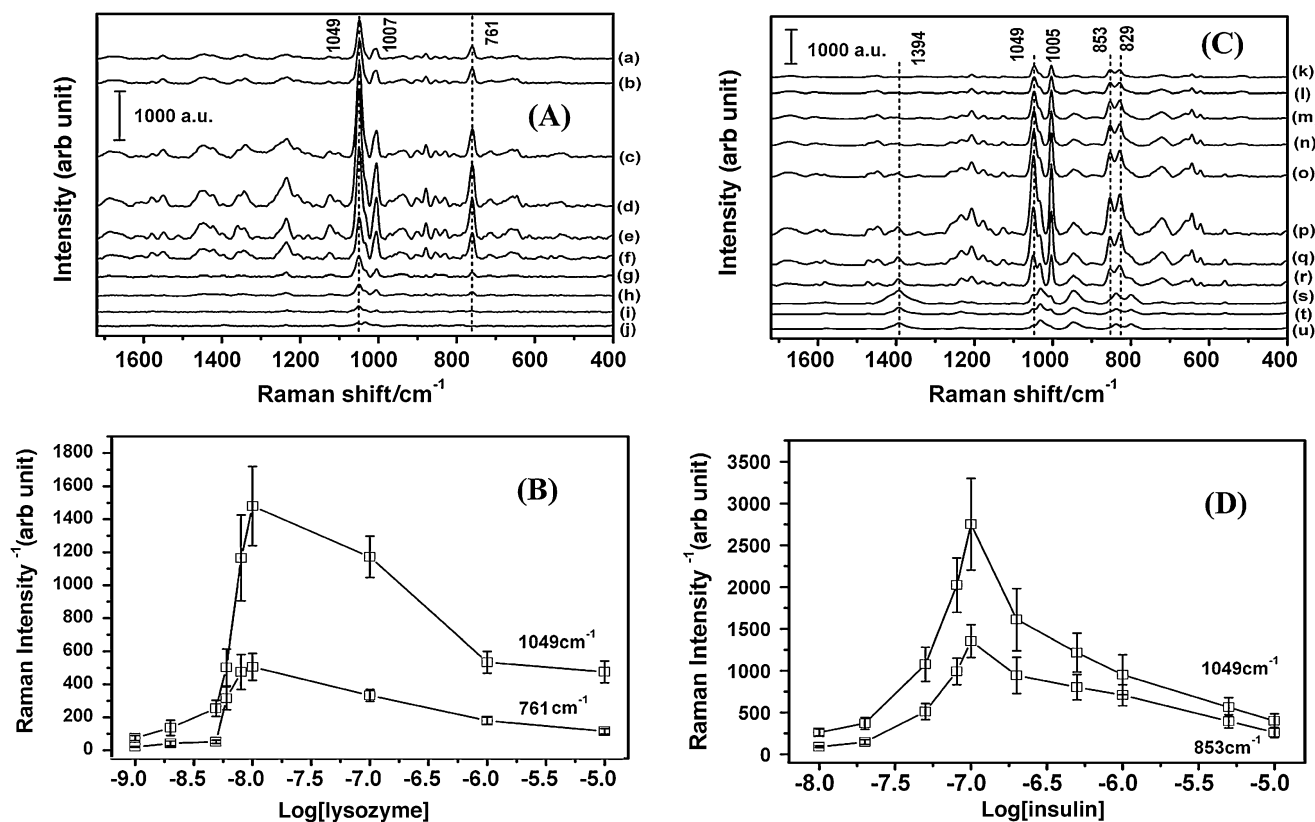


**Figure 2.** Normal Raman spectra of solid (a) lysozyme and (d) insulin, heat-induced SERS spectra of  $10^{-5}$  M (b) lysozyme and (c) insulin.

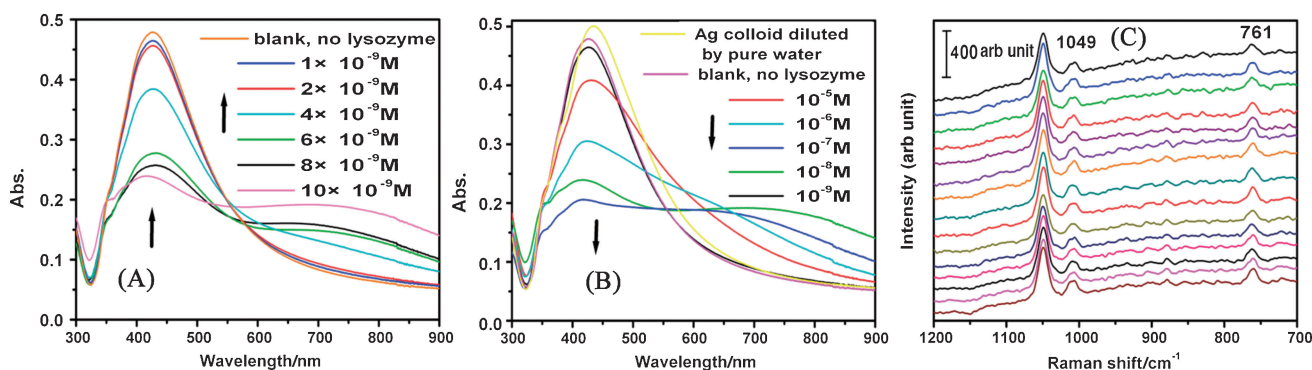
spectra are shown in Figure 2, and these band assignments are summarized in Table 1 of SI.<sup>22</sup> Compared to the corresponding intensities in the SERS spectrum of insulin, the band intensities of Tyr residue (1618, 1207, 1177, 853, and 829  $\text{cm}^{-1}$ ) in SERS spectrum of lysozyme are stronger while the band intensities of Trp residue (1548, 1359, 1340, 878, and 761  $\text{cm}^{-1}$ ) and disulfide ( $\nu(\text{S-S})$  and/or  $\nu(\text{C-S})$ , 520–750  $\text{cm}^{-1}$ ) are weaker. This trend is consistent with the appearance probabilities of the residues (see

Table 2 of SI),<sup>22</sup> further confirming the band assignments for the SERS spectra.

The intensity and sharpness of the  $\text{NO}_3^-$  peak at 1049  $\text{cm}^{-1}$  lead us to propose that this band may be suitable for quantitative analysis of the proteins. Figure 3A displays concentration-dependent ( $10^{-5}$ – $10^{-9}$  M) SERS spectra of lysozyme. Comparison to previous studies, the results in Figure 3A demonstrate two novelties. First, the detection limit is much lower: down to a submonolayer coverage of  $10^{-9}$  M (the coverage calculation is reported in SI<sup>22</sup>), which is more sensitive than the method proposed before by which the detection limit is  $10^{-6}$   $\text{g L}^{-1}$  (equivalent to  $7 \times 10^{-8}$  M).<sup>6</sup> Second, a concentration-dependent curve shows a bell shape rather than a sigmoidal one.<sup>13</sup> Similar results are displayed in Figures 3C and 3D for insulin. The bell shape result for lysozyme (Figure 3B) is supported by UV-visible spectra of mixtures of lysozyme and Ag colloids shown in Figures 4A and 4B. Figure 4A shows that the absorption of UV-visible spectrum decreases with the protein in the range of  $10^{-8}$ – $10^{-9}$  M. On the other hand, in the range of  $10^{-5}$ – $10^{-9}$  M (Figure 4B), there is a minimum absorption at  $10^{-7}$  M, indicating that when the protein concentration is above monolayer coverage, a further increase in protein concentration would cause a negative effect against silver aggregation and decrease the SERS signal. This phenomenon was not found when glutathione was used as an analyte.<sup>13</sup> The difference leads to the proposal that the crowding of the silver colloid environment is respon-



**Figure 3.** (A) Concentration-dependent ( $10^{-5}$ – $10^{-9}$  M) SERS spectra of lysozyme (a)  $1 \times 10^{-5}$ , (b)  $1 \times 10^{-6}$ , (c)  $1 \times 10^{-7}$ , (d)  $1 \times 10^{-8}$ , (e)  $8 \times 10^{-9}$ , (f)  $6 \times 10^{-9}$ , (g)  $4 \times 10^{-9}$ , (h)  $2 \times 10^{-9}$ , (i)  $1 \times 10^{-9}$ , and (j)  $1 \times 10^{-10}$  M; (B) Raman intensity at 1049 ( $\text{NO}_3^-$  band) and 761  $\text{cm}^{-1}$  (Tyr band) of the SERS spectra in (A); (C) concentration-dependent ( $10^{-5}$ – $10^{-8}$  M) SERS spectra of insulin (k)  $1 \times 10^{-5}$ , (l)  $5 \times 10^{-6}$ , (m)  $1 \times 10^{-6}$ , (n)  $5 \times 10^{-7}$ , (o)  $2 \times 10^{-7}$ , (p)  $1 \times 10^{-7}$ , (q)  $8 \times 10^{-8}$ , (r)  $5 \times 10^{-8}$ , (s)  $2 \times 10^{-8}$ , and (t)  $1 \times 10^{-9}$  M; (u) 0 M control; (D) Raman intensity at 1049 and 853  $\text{cm}^{-1}$  (Trp band) of the SERS spectra in (C).



**Figure 4.** UV absorption spectra of silver colloid measured at 5 min after addition of lysozyme solutions. The final concentrations of lysozyme in silver colloid solutions were (A)  $10^{-8}$ – $10^{-9}$  and (B)  $10^{-5}$ – $10^{-9}$  M; (C) a collection of 15 SERS spectra of lysozyme ( $10^{-5}$  M), measured at 15 different points of the sample and averaged to obtain the spectrum (c) in Figure 1. The standard deviation of the spectra intensities at  $1049\text{ cm}^{-1}$  is 7.5%.

sible for the silver colloid aggregation. As lysozyme is a large and hard molecule<sup>16</sup> (129 amino acids) compared to the small and soft glutathione (3 amino acids), it is reasonable to observe that concentrated lysozyme inhibits silver colloid aggregation while the small peptide does not.

Nonlinear curves for quantitative or semiquantitative SERS studies have been found in previous studies.<sup>17–19</sup> In the present work, it shows a bell shape, which means that there are two possible concentrations for a certain SERS intensity. However, dilution of the protein sample can solve the problem.  $[C_{\text{max}}]$  represents the protein concentration corresponding to the maximum SERS intensity. Dilution of the protein sample with concentration higher than  $[C_{\text{max}}]$  can induce an increase in SERS intensity while dilution of that with concentration lower than  $[C_{\text{max}}]$  leads to a decrease in SERS intensity.

The adsorption mechanism of protein and  $\text{NO}_3^-$  possibly follows the most common co-adsorption scenario, which has been well established as a double-layer adsorption rule.<sup>20</sup> Namely, there is a layer of electrolyte such as citrate and  $\text{NO}_3^-$  between an adsorbed layer of proteins and Ag colloid. The results in Figure 3 provide some support to this speculation, suggesting that the intensities of protein bands including those at  $761$  and  $853\text{ cm}^{-1}$  are positively correlated with the intensity of the  $\text{NO}_3^-$  band, which may be evidence for the co-adsorption manner of the protein and  $\text{NO}_3^-$ . A detailed depiction of the adsorption mechanism of protein and electrolyte was published recently.<sup>21</sup>

Peaks due to acid radicals are usually considered undesirable background in SERS studies, and additional caution should be paid to avoid these peaks. Here, however, we have developed a novel method that can selectively enhance the peak at  $1049\text{ cm}^{-1}$  originating from  $\text{NO}_3^-$  to make it valuable for semiquantitative detection of proteins without any additional resonant effect. This method is simple, rapid (drying time is 210 s), reproducible (Figure 4C), and label-free. The laser power is low (6 mW) and the exposure time is short (20 s), which meets the need for routine analyses. The intensity and the sharpness of the  $\text{NO}_3^-$  peak suggest its potential in more extended applications, especially those to biomolecules with low Raman cross sections and poor quality spectra by normal SERS protocols. Also, the bell shape indicates the potential of this method in exploring adsorption phenomenon of a protein on a colloidal interface. These studies are ongoing in our group.<sup>22</sup>

#### References and Notes

- 1 *Surface-Enhanced Raman Scattering: Physics and Applications*, ed. by K. Kneipp, M. Moskovits, H. Kneipp, Springer, Berlin, Germany, **2006**.
- 2 R. Aroca, *Surface-Enhanced Vibrational Spectroscopy*, John Wiley & Sons, Ltd., Chichester, UK, **2006**.
- 3 T. Itoh, A. Sujith, Y. Ozaki, in *Frontiers of Molecular Spectroscopy*, ed. by J. Laane, Elsevier, Amsterdam, **2009**.
- 4 S. E. J. Bell, N. M. S. Sirimuthu, *J. Am. Chem. Soc.* **2006**, *128*, 15580.
- 5 Y. Wang, H. Wei, B. Li, W. Ren, S. Guo, S. Dong, E. Wang, *Chem. Commun.* **2007**, 5220.
- 6 X. X. Han, G. G. Huang, B. Zhao, Y. Ozaki, *Anal. Chem.* **2009**, *81*, 3329.
- 7 S. Stewart, P. M. Fredericks, *Spectrochim. Acta, Part A* **1999**, *55*, 1615.
- 8 V. P. Drachev, M. D. Thoreson, E. N. Khaliullin, V. J. Davisson, V. M. Shalaev, *J. Phys. Chem. B* **2004**, *108*, 18046.
- 9 G. Das, F. Mecarini, F. Gentile, F. de Angelis, M. Kumar, P. Candeloro, C. Liberale, G. Cuda, E. di Fabrizio, *Biosens. Bioelectron.* **2009**, *24*, 1693.
- 10 G. V. P. Kumar, R. Selvi, A. H. Kishore, T. Kundu, C. Narayana, *J. Phys. Chem. B* **2008**, *112*, 6703.
- 11 I. Pavel, E. McCarney, A. Elkhaled, A. Morrill, K. Plaxco, M. Moskovits, *J. Phys. Chem. C* **2008**, *112*, 4880.
- 12 E. Podstawka, Y. Ozaki, L. M. Proniewicz, *Appl. Spectrosc.* **2004**, *58*, 1147.
- 13 G. G. Huang, X. X. Han, M. K. Hossain, Y. Ozaki, *Anal. Chem.* **2009**, *81*, 5881.
- 14 C. H. Munro, W. E. Smith, M. Garner, J. Clarkson, P. C. White, *Langmuir* **1995**, *11*, 3712.
- 15 M. F. Mrozek, M. Weaver, *Anal. Chem.* **2002**, *74*, 4069.
- 16 C. A. Haynes, W. Norde, *J. Colloid Interface Sci.* **1995**, *169*, 313.
- 17 I. T. Shadi, B. Z. Chowdhry, M. J. Snowden, R. Withnall, *Appl. Spectrosc.* **2000**, *54*, 384.
- 18 S. E. J. Bell, N. M. S. Sirimuthu, *Chem. Soc. Rev.* **2008**, *37*, 1012.
- 19 C. McLaughlin, D. MacMillan, C. McCardle, W. E. Smith, *Anal. Chem.* **2002**, *74*, 3160.
- 20 C. A. Haynes, W. Norde, *Colloids Surf., B* **1994**, *2*, 517.
- 21 W. R. Glomm, Ø. Halskau, A.-M. D. Hanneseth, S. Volden, *J. Phys. Chem. B* **2007**, *111*, 14329.
- 22 Supporting Information is available electronically on the CSJ-Journal Web site, <http://www.csj.jp/journals/chem-lett/index.html>.